JAYPEE UNIVERSITY OF INFORMATION TECHNOLOGY, WAKNAGHAT TEST -2 EXAMINATION- 2023

BTech-V Semester (CSE/IT)

COURSE CODE (CREDITS): 18B1WPH532 (3) COURSE NAME: Applied Materials Science

COURSE INSTRUCTORS: PBB, VSA, SKT, HAZ

MAX. MARKS: 25

MAX. TIME: 1.5 H

Note: (a) All questions are compulsory.

(b) Marks are indicated against each question in square brackets.

(c) The candidate is allowed to make suitable numeric assumptions wherever required for solving problems

Q1. (a) Discuss the phenomenon of poling in ceramic processing.

(b) For an atom subjected to externally applied magnetic field, derive and expression for Larmour precession frequency and discuss the result. [3-marks] [CO-2]

(c) A paramagnetic material has 10^{28} atoms/m³. The magnetic moment of each atom is 1.8×10^{-23} Am². Calculate the paramagnetic susceptibility at room temperature (~300 K). [2-marks] [CO-2]

Q2. (a) Write a short note on ferroelectric domains.

[2-marks] [CO-5]

(b) Plot the graphs for Curie's law and Curie-Weiss law for a paramagnetic sample and discuss the differences.

[3-marks] [CO-2]

(c) Consider a helium atom in its ground state (1s). The mean radius in the Langevin formula may be approximated by Bohr radius 0.529Å. The density of helium is 0.178 kg/m³. Calculate the diamagnetic susceptibility of a helium atom. [3-marks] [CO-2]

Q3. (a) Discuss the stress-strain diagram for a metal. Also explain strain-hardening.

[3-marks] [CO-5]

(b) If a dielectric sample has its dielectric constant 4 and refractive index 1.5, then calculate the ratio of ionic to electronic polarizabilities: [3-marks] [CO-3]

Q4. (a) Sulphur has atomic weight 32 and its density is 2.08 g/cc. The electronic polarizability of sulphur is 3.5x10⁻⁴⁰ Fm². Considering the cubic symmetry in sulphur solid, calculate its relative dielectric constant.

[2-marks] [CO-3]

(b) Using atomic theory, discuss the origin of magnetic moment and the subsequent quantization.

[2-marks] [CO-1]