## JAYPEE UNIVERSITY OF INFORMATION TECHNOLOGY, WAKNAGHAT TEST -3 EXAMINATION- 2024

## B.Tech-III Semester (BT/BI)

COURSE CODE (CREDITS): 18B11BT314 (3)

MAX. MARKS: 35

COURSE NAME: GENERAL CHEMISTRY

COURSE INSTRUCTORS: Dr. Gopal Singh Bisht

MAX. TIME: Hours

Note: (a) All questions are compulsory.

(b) The candidate is allowed to make Suitable numeric assumptions wherever required for solving problems

| student prepared a mixture of L-alanine and its enantiomer and 7 gm of the mixture was dissolved in 20 mL of water. The solution was then placed in a sample cell with pathlength of 10 cm. and observed optical rotation is +0.8 Calculate % ee of mixture.  b) A solution of ethanol in water is 15% by volume. If the solution and pure ethanol have densities of 0.9066g/cc and 0.785 g/cc respectively. Find % of ethanol by weight.  c) Explain principle of IR spectroscopy.  | COI :   | Marks 2 2 |
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|  |         | 1         |
| O2 a) Examine a polymer from its chemical structure. CO  | 20 V 3  |           |
|  |         |           |
| stereochemistry or application. Give suitable examples.  |         |           |
| b) Consider the following feactions. Predict whether an increase   1   | coil 2  | ,         |
| in temp will favor reactant or product. Justify your prediction  | 2       | r         |
| Catalad  |         |           |
| + Hg Catalyt   |         |           |
| d d  |         |           |
| At the second se |         |           |
| Give JUPAC names to following organic compound (with   | 011     |           |
| assigning proper stereochemistry)  | -       |           |
| O RY   |         |           |
|  |         |           |
| H Y  |         |           |
| HQ 0   |         |           |
| (b) \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \  |         |           |
|  |         |           |
| 23. a) Design a synthetic route to prepare dipeptides Phe-Ala using CC   | OIV 4   |           |
| appropriate reagents. Examine the challenges in the synthesis  | ~ T     |           |
| and how to tackle them. (Give Labeled reactions)   |         |           |
|  | O IV 2. | .5        |

| b) A peptide containing equimolar amounts of met + phe +asp ser + thr was treated with CNBr. A peptide and a single amin acid (identified as homoserine) were released. Treatment of the original pentapeptide with chymotrypsin produced two fragments, one of which was significantly more acidic than the other. The acidic fragment contained methionine. Treatment of the original pentapeptide with carboxypeptidase A yielded serine very rapidly, followed by threonine. Deduce the amino acid sequence of the pentapeptide. | o e e e f di CO IV | 2.5  |
|--|--------------------|------|
| c) Propose two different routes for synthesis of racemic mixture of Phenylalanine using appropriate reagents.  |                    | Mari |
| Q4. If the compounds below undergoes elimination reactions, predict their  | 1                  | ļ    |
| preferred mechanism and justify your answer.   | Com                | 3    |
| $\begin{array}{c ccccccccccccccccccccccccccccccccccc$  |                    |      |
| $(C) \qquad \begin{array}{c} X \\ X \\ CH_3CO CH_2-CH-CH_3 \end{array}$  |                    |      |
| Q5. a) Examine the structural features of simple lipids such as steroid,   |                    |      |
| prostaglandin and terpenes.  | COIV               | 3    |
| b) Draw a mechanism for the base catalyzed hydrolysis in following reactions.  | COIV               | 2    |
| CH <sub>2</sub> -OH  |                    | 1    |
| CH <sub>2</sub> -O-C-(CH <sub>2</sub> ) <sub>16</sub> CH <sub>3</sub> CH-OH  | 1                  |      |
| CH2-OH   |                    |      |
| CH-O-C-(CH <sub>2</sub> ) <sub>16</sub> CH <sub>3</sub> + NaOH -   |                    |      |
| $CH_2-O-C-(CH_2)_{16}CH_3$ + 3 Na <sup>+</sup> O-C-(CH <sub>2</sub> ) <sub>16</sub> CH <sub>3</sub>  |                    |      |
| Q6. a) When a diglucose is treated with ethanol in the presence of an  | COV                | 2.5  |
| acid catalyst, such as HCl, two products are formed. Draw both   |                    | 2.3  |
| products and account for their formation with a mechanism  |                    |      |
| Draw proper structure to show hydrogen-bonding interactions  | cov                | 2    |
| that occur between complementary base pairs in DNA.  Compare the structure of starch, cellulose and glycogens.   | 00.11              |      |
| staten, centiose and glycogens.  | CO V               | 1.5  |