

JAYPEE UNIVERSITY OF INFORMATION TECHNOLOGY, WAKNAGHAT  
MAKE UP EXAMINATION NOV - 2025

B.Tech-V Semester (CSE/IT)

COURSE CODE (CREDITS): 18B1WPH532 (03)

MAX. MARKS: 25

COURSE NAME: APPLIED MATERIALS SCIENCE

COURSE INSTRUCTORS: PBB, VSA, SKT, SBA, HAZ

MAX. TIME: 1 Hour 30 minutes

**Note:** (a) All questions are compulsory.

(b) The candidate is allowed to make Suitable numeric assumptions wherever required for solving problems. Scientific Calculators are allowed.

Q.No	Question	CO	Marks										
Q1	(a) Calculate the percentage ionic polarizability in sodium chloride which has the refractive index and static dielectric constant 1.5 and 5.0 respectively.	3	3										
	(b) Discuss the role of porosity with respect to the properties in ceramic materials.	5	2										
Q2	(a) What is dielectric loss? Obtain the expression for energy loss in a dielectric material.	1	3										
	(b) Calculate the percentage increase in magnetic induction when the space within a current carrying toroid is filled with magnesium with susceptibility $1.2 \times 10^{-5}$ .	3	2										
Q3	(a) Obtain the expression of classical paramagnetic susceptibility and discuss its variation at low and high temperatures.	2	4										
	(b) The saturation value of magnetization of iron is $1.74 \times 10^6$ A/m. Iron has body centered cubic structure with an elementary cube edge of 0.286 nm. Calculate the average number of Bohr magnetons contributed to the magnetization per atom.	3	2										
Q4	(a) Show that for simple two sublattice model, the neel temperature and the curie temperature are same for an antiferromagnetic material.	2	3										
	(b) Plot the variation of inverse of susceptibility with temperature for para-, ferro- and antiferro- magnetic materials in a single graph.	1	2										
Q5	Determine degree of polymerization and polydispersity index for following Polyethylene $-(C_2H_4)_n-$ sample:	3	4										
	<table border="1"> <tr> <td><math>M_i</math> (g/mol)</td> <td>2600</td> <td>7800</td> <td>13000</td> <td>18000</td> </tr> <tr> <td><math>N_i</math></td> <td>4200</td> <td>8050</td> <td>7100</td> <td>2050</td> </tr> </table>	$M_i$ (g/mol)	2600	7800	13000	18000	$N_i$	4200	8050	7100	2050		
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<p>Constants: <math>m_e = 9.11 \times 10^{-31}</math> kg; <math>e = 1.6 \times 10^{-19}</math> C; <math>N_A = 6.023 \times 10^{23}</math>; <math>k_B = 1.38 \times 10^{-23}</math> J/K;  <math>\epsilon_0 = 8.85 \times 10^{-12}</math> F/m; <math>\mu_B = 9.27 \times 10^{-24}</math> Am<sup>2</sup>;</p>													