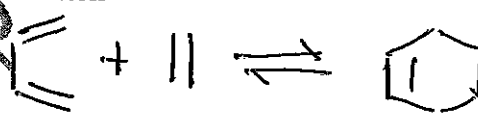
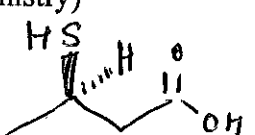
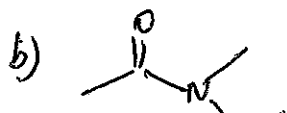
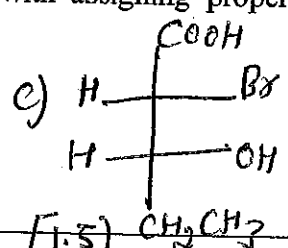


Note: (a) All questions are compulsory.

(b) The candidate is allowed to make Suitable numeric assumptions wherever required for solving problems

| Q.No | Question | CO | Marks |
|------|--|------|--|
| Q1 | <p>a) The specific rotation of L-alanine in water (25°C) is +2.8. A student prepared a mixture of L-alanine and its enantiomer and 7 gm of the mixture was dissolved in 20 mL of water. The solution was then placed in a sample cell with pathlength of 10 cm. and observed optical rotation is +0.8. Calculate % ee of mixture.</p> <p>b) A solution of ethanol in water is 15% by volume. If the solution and pure ethanol have densities of 0.9066g/cc and 0.785 g/cc respectively. Find % of ethanol by weight.</p> <p>c) A weak acid has a pK_a of 7.07. If the pH of the solution is 7.38, what percentage of the acid is undissociated?</p> <p>d) A sample of water contains 6.585×10^{20} atoms. How much water (in mol) is present?</p> <p>e) 200 gm of HCl solution of relative density 1.17 contains 66.8 gm of HCl. What volume of this HCl solution will be required to exactly neutralize 5 L of 0.1 N NaOH.</p> | COI | <p>[2]</p> <p>[2]</p> <p>[2]</p> <p>[1]</p> <p>[2]</p> |
| Q2 | Design an experiment to determine whether a given nucleophilic substitution reaction follows an S_N1 or S_N2 mechanism. Assess the steps, variables to control, and the expected outcomes that would indicate the mechanism. | COII | [5] |
| Q3 | Consider the following reactions. Predict whether an increase in temp will favor reactant or product. Justify your prediction | COII | [3] |
| |  | | |
| Q4 | <p>a) If the compounds undergoes elimination reactions, explain possibility of preferred mechanism and justify your answer.</p> <p>b) Explain Miller indices with suitable example.</p> | COII | <p>[3]</p> <p>[2]</p> |
| Q5 | Give IUPAC names to following organic compound (with assigning proper stereochemistry) | COII | [3] |
| | <p>a)  [1]</p> <p>b)  [0.5]</p> <p>c)  [1.5]</p> | | |