

JAYPEE UNIVERSITY OF INFORMATION TECHNOLOGY, WAKNAGHAT

TEST -3 EXAMINATION - 2016

M.Tech I Semester

COURSE CODE: 14M31CE212

MAX. MARKS: 35

COURSE NAME: CONTAMINANT TRANSPORT

COURSE CREDITS: 03

MAX. TIME: 2 Hr

Note: All questions are compulsory. Carrying of mobile phone during examinations will be treated as case of unfair means.

- Q1.a) Distinguish between "Direct" and "Indirect" photodegradation. Explain the two processes by giving examples [2.5 Marks]
- b) What is the light intensity at 1-m depth in a lake, given an intensity of $3000 \mu\text{E}/(\text{m}^2 \cdot \text{sec})$ just beneath the lake's surface and an extinction coefficient (base e) of 0.6/m? If an aquatic plant has a light compensation point (the light intensity at which respiration rate equals photosynthetic rate) of $150 \mu\text{E}/(\text{m}^2 \cdot \text{sec})$, what is the maximum depth at which the plant may be expected to grow? [2.5 Marks]
- Q2.a) Discuss why hydrolysis reaction are important and significant processes in the aquatic environment [2.5 Marks]
- b) What are the important parameters that affect the rates of hydrolysis in the environment? [2.5 Marks]
- Q3.a) Define "Retardation Factor". Derive the relationship between retardation factor, distribution coefficient, bulk density and porosity. [1.5 + 2.0 Marks]
- b) The porous material of an aquifer has a bulk density of $2 \text{g}/\text{cm}^3$ and contains 0.5% organic carbon. Estimate the retardation factor for naphthalene (C_{10}H_8). If the porosity of the aquifer is 0.24, the hydraulic conductivity is $10^{-3} \text{cm}/\text{sec}$, and the hydraulic gradient is 0.001, how fast will a plume of naphthalene advance in the aquifer? $\log K_{oc} = 0.937 \log K_{ow} - 0.006$; $\log K_{ow} = 3.36$ [04 Marks]
- Q4.a) What are the dry deposition mechanisms by which physical removal of particles takes place in atmosphere? Discuss on any one mechanism. [2.5 Marks]
- b) An underground nuclear bomb test results in deposition of particles into the atmosphere at a height of 1.5km over the site. If the soil particles have a density of $2.3 \text{g}/\text{cm}^3$, and the wind speed is 1.5 m/sec, what would be the maximum size of the particles that could be transported beyond the border of the test site, which is 200km downwind? Assume dynamic viscosity of air = $1.83 \times 10^{-4} \text{g}/(\text{cm} \cdot \text{sec})$ [2.5 Marks]

- Q5.a) What are the two major differences between settling of particles in water and air? **[02 Marks]**
- b) Estimate the dry deposition rate of nitric acid (HNO_3) aerosol on a forest, if atmospheric concentration of HNO_3 is $5\mu\text{g}/\text{m}^3$. Assume a deposition velocity of 1 cm/sec. **[02 Marks]**
- c) What should be the maximum radii of a particle in air to which stokes's law is applicable **[01 Marks]**
- Q6.a) How particulate chemicals are removed from atmosphere through wet deposition processes. **[02 Marks]**
- b) With the help of a neat figure, explain the working of a wet-dry atmospheric deposition collector **[03 Marks]**
- c) Write a note on Reactive Oxygen Species (ROS) **[2.5 Marks]**